

Aphanamixoid A, a Potent Defensive Limonoid, with a New Carbon Skeleton from *Aphanamixis polystachya*

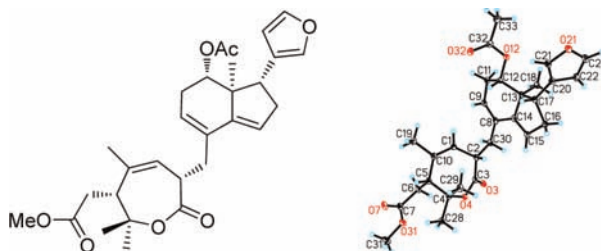
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ABSTRACT



Aphanamixoid A (1), a limonoid with a new carbon skeleton, along with its biogenetically related limonoid aphanamixoid B (2), was isolated from the leaves and twigs of *Aphanamixis polystachya*. Their structures with the absolute stereochemistry were determined by spectroscopic analysis, X-ray crystallography and computational methods. The significant antifeedant activity of 1 against the generalist plant-feeding insect *Helicoverpa armigera* ($EC_{50} = 0.015 \mu\text{mol}/\text{cm}^2$) suggested it may be a potent defensive component of *A. polystachya*.

Limonoids, a series of structurally diverse and highly oxygenated tetranortriterpenoids mainly found in the family of Meliaceae, have been attracting continuous attention from biogenetic and synthetic points of view.¹ In recent years, a number of limonoids have still been

isolated by several research groups, a few of which exhibited biological activities including cytotoxic,² antimalarial,³ insect antifeedant,⁴ insecticidal,^{4a,5} and insect growth regulatory^{4b} activities.

The plant *Aphanamixis polystachya* (Wall.) R. N. Parker (Meliaceae), a timber tree, is mainly distributed in the tropical areas of Asia, such as India, Malaysia, Indonesia, and southern China.⁶ Previous chemical studies on this

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(1) (a) Fang, X.; Di, Y. T.; Hao, X. J. *Curr. Org. Chem.* **2011**, *15*, 1363–1391. (b) Tan, Q. G.; Luo, X. D. *Chem. Rev.* **2011**, *111*, 7437–7522. (c) Fang, X.; Di, Y. T.; He, H. P.; Liu, H. Y.; Zhang, Z.; Ren, Y. L.; Gao, Z. L.; Gao, S.; Hao, X. J. *Org. Lett.* **2008**, *10*, 1905–1908. (d) Han, M. L.; Zhang, H.; Yang, S. P.; Yue, J. M. *Org. Lett.* **2012**, *14*, 486–489. (e) Fukuzaki, T.; Kobayashi, S.; Hibi, T.; Ikuma, Y.; Ishihara, J.; Kanoh, N.; Murai, A. *Org. Lett.* **2002**, *4*, 2877–2880. (f) Trudeau, S.; Morken, J. P. *Org. Lett.* **2005**, *7*, 5465–5468. (g) Zhang, Q.; Di, Y. T.; He, H. P.; Fang, X.; Chen, D. L.; Yan, X. H.; Zhu, F.; Yang, T. Q.; Liu, L. L.; Hao, X. J. *J. Nat. Prod.* **2011**, *74*, 152–157. (h) Yin, J. L.; Di, Y. T.; Fang, X.; Liu, E. D.; Liu, H. Y.; He, H. P.; Li, S. L.; Li, S. F.; Hao, X. J. *Tetrahedron Lett.* **2011**, *52*, 3083–3085.

(2) (a) Wang, X. N.; Yin, S.; Fan, C. Q.; Wang, F. D.; Lin, L. P.; Ding, J.; Yue, J. M. *Org. Lett.* **2006**, *8*, 3845–3848. (b) Murphy, B. T.; Brodie, P.; Sleboadnick, C.; Miller, J. S.; Birkinshaw, C.; Randrianjanaka, L. M.; Andriantsiferana, R.; Rasamison, V. E.; TenDyke, K.; Suh, E. M.; Kingston, D. G. I. *J. Nat. Prod.* **2008**, *71*, 325–329. (c) Ning, J.; Di, Y. T.; Fang, X.; He, H. P.; Wang, Y. Y.; Li, Y.; Li, S. L.; Hao, X. J. *J. Nat. Prod.* **2010**, *73*, 1327–1331.

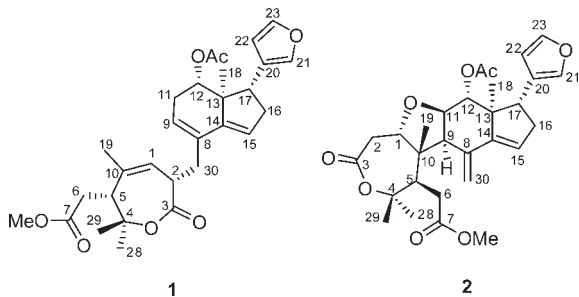
(3) (a) Yin, S.; Wang, X. N.; Fan, C. Q.; Liao, S. G.; Yue, J. M. *Org. Lett.* **2007**, *9*, 2353–2356. (b) Hay, A. E.; Ioset, J. R.; Ahua, K. M.; Diallo, D.; Brun, R.; Hostettmann, K. *J. Nat. Prod.* **2007**, *70*, 9–13.

(4) (a) Isman, M. B.; Koul, O.; Luczynski, A.; Kaminski, J. *J. Agric. Food Chem.* **1990**, *38*, 1406–1411. (b) Fowles, R.; Mootoo, B.; Ramsewak, R.; Khan, A.; Ramsubhag, A.; Reynolds, W.; Nair, M. *Pest Manage. Sci.* **2010**, *66*, 1298–1303. (c) Nakatani, M.; Abdelgaleil, S. A. M.; Kurawaki, J.; Okamura, H.; Iwagawa, T.; Doe, M. *J. Nat. Prod.* **2001**, *64*, 1261–1265. (d) Koul, O.; Daniewski, W. M.; Multani, J. S.; Gumulka, M.; Singh, G. *J. Agric. Food Chem.* **2003**, *51*, 7271–7275.

(5) (a) Xu, H.; Zhang, J. L. *Bioorg. Med. Chem. Lett.* **2011**, *21*, 1974–1977. (b) Ge, Y. H.; Zhang, J. X.; Mu, S. Z.; Chen, Y.; Yang, F. M.; Lu, Y.; Hao, X. J. *Tetrahedron* **2012**, *68*, 566–572.

(6) Chen, S. K.; Chen, B. Y.; Li, H. *Flora Reipublicae Popularis Sinicae (Zhongguo Zhiwu Zhi)*; Science Press: Beijing, China, 1997; Vol. 43, pp 239–240.

plant have resulted in the isolation of a few new limonoids; however, no significant bioactivity has been found from those compounds.⁷ In the current study, a new limonoid with potent antifeedent activity, aphanamixoid A (**1**), was isolated from the leaves and twigs of *A. polystachya* collected in the Yunnan province of China. **1** could be derived from a new biogenetically related limonoid, aphanamixoid B (**2**), via the unique cleavage of a C-9–C-10 bond as well as the formation of a C-2–C-30 bond by means of 3,3-rearrangement. In this paper, we report the isolation, structure elucidation, plausible biogenetic pathway, and the bioactivities of aphanamixoids A (**1**) and B (**2**).



Aphanamixoid A (**1**)⁸ was obtained as colorless crystals (in acetone). Its molecular formula, C₂₉H₃₆O₇, was established from the quasi-molecular ion peak at *m/z* 519.2361 [M + Na]⁺ (calcd 519.2358, C₂₉H₃₆O₇Na) in the positive HRESIMS, which indicated 12 degrees of unsaturation. UV absorption at 242 nm (3.44) indicated the presence of conjugated double bonds. IR peaks at 1732 and 1717 cm⁻¹ as well as ¹³C NMR signals at δ 173.4, 172.7, and 170.9 revealed three ester carbonyl groups. Besides a methoxy group (δ_{H} 3.70; δ_{C} 52.1) and an acetyl group (δ_{H} 1.90; δ_{C} 21.3, 170.9), **1** contained 26 carbons, including a β -furan ring (δ_{H} 6.32, 7.23, 7.35; δ_{C} 111.6, 140.0, 142.1) and four tertiary methyl groups (δ_{H} 0.90, 1.36, 1.61, 1.78). The above evidence suggested that **1** was a tetranortriterpenoid. Furthermore, apart from the five double bonds and three carbonyl groups, the remaining four degrees of unsaturation indicated **1** to be tetracyclic system.

Extensive comparison of ¹H and ¹³C NMR data with those of a known limonoid, munronoid B, suggested both

(7) (a) Yang, S. P.; Chen, H. D.; Liao, S. G.; Xie, B. J.; Miao, Z. H.; Yue, J. M. *Org. Lett.* **2011**, *13*, 150–153. (b) Zhang, Y.; Wang, J. S.; Wang, X. B.; Wei, D. D.; Luo, J. G.; Luo, J.; Yang, M. H.; Kong, L. Y. *Tetrahedron Lett.* **2011**, *52*, 2590–2593. (c) Zhang, H.; Chen, F.; Wang, X.; Wu, D.; Chen, Q. *Magn. Reson. Chem.* **2007**, *45*, 189–192. (d) Sadhu, S. K.; Phattanawasin, P.; Choudhuri, M. S. K.; Ohtsuki, T.; Ishibashi, M. *J. Nat. Med.* **2006**, *60*, 258–260. (e) Mulholland, D. A.; Naidoo, N. *Phytochemistry* **1999**, *51*, 927–930.

(8) Aphanamixoid A (**1**): colorless crystals (in acetone); mp 148–150 °C; HRESIMS at *m/z* 519.2361 [M + Na]⁺ (calcd 519.2358, C₂₉H₃₆O₇Na); [α_{D}^{20}] = +39.9° (*c* 0.260, MeOH); UV (MeOH) λ_{max} (log ϵ) 242 (3.44), 202 (3.39); CD (0.000511 M, MeOH) λ_{max} ($\Delta\epsilon$) 231 (1.14); IR ν_{max} (KBr) cm⁻¹ 1732, 1717, 1265, 1255, and 1245; ¹H and ¹³C NMR data, see Table 1.

(9) Flack, H. D. *Acta Crystallogr., Sect. A* **1983**, *39*, 876–881.

(10) Aphanamixoid B (**2**): colorless amorphous powder; mp 170–171 °C; HRESIMS at *m/z* 535.2314 [M + Na]⁺ (calcd 535.2307, C₂₉H₃₆O₈Na); [α_{D}^{20}] = +81.8° (*c* 0.140, MeOH); UV (MeOH) λ_{max} (log ϵ) 210 (3.06), 202 (3.07); CD (0.000342 M, MeOH) λ_{max} ($\Delta\epsilon$) 236 (1.81), 205 (0.56); IR (KBr) ν_{max} 2957, 2930, 1739, 1373, 1235, and 1030 cm⁻¹; ¹H and ¹³C NMR data, see Table 1.

Table 1. ¹H and ¹³C NMR Data for **1** and **2** in CDCl₃

	1		2	
	$\delta_{\text{C}}^{\text{a}}$	δ_{H} (mult; <i>J</i> , Hz) ^a	$\delta_{\text{C}}^{\text{a}}$	δ_{H} (mult; <i>J</i> , Hz) ^b
1	120.2 (d)	5.18 (s)	81.6 (d)	4.07 (dd, 10.9, 4.8)
2	43.7 (d)	3.42 (br s)	38.2 (t)	2.87 (m, 2H)
3	172.7 (s)		170.3 (s)	
4	81.4 (s)		84.8 (s)	
5	47.8 (d)	2.74 (dd, 6.2, 4.0)	48.6 (d)	2.90 (dd, 6.9, 3.8)
6a	34.4 (t)	2.23 (dd, 17.0, 4.0)	33.4 (t)	2.41 (dd, 17.0, 6.9)
6b		2.82 (dd, 17.0, 6.2)		3.07 (dd, 17.0, 3.8)
7	173.4 (s)		173.7 (s)	
8	131.0 (s)		140.3 (s)	
9	125.0 (d)	5.50 (d, 3.4)	55.6 (d)	3.06 (d, 8.2)
10	137.8 (s)		50.5 (s)	
11 α	29.9 (t)	2.22 (m)	79.3 (d)	4.11 (dd, 9.7, 8.2)
11 β		2.47 (m)		
12	77.6 (d)	5.07 (dd, 10.0, 6.1)	75.8 (d)	5.35 (d, 9.7)
13	50.2 (s)		51.7 (s)	
14	147.6 (s)		149.4 (s)	
15	121.7 (d)	5.65 (s)	122.9 (d)	5.76 (t, 3.0)
16 α	37.7 (t)	2.58 (m, 2H)	37.8 (t)	2.39 (ddd, 16.5, 10.5, 3.0)
16 β				2.62 (ddd, 16.5, 8.4, 3.0)
17	46.4 (d)	3.15 (dd, 9.9, 8.7)	47.2 (d)	3.30 (dd, 10.5, 8.4)
18	13.2 (q)	0.90 (s)	15.1 (q)	0.80 (s, CH ₃)
19	25.5 (q)	1.78 (s)	20.4 (q)	1.17 (s, CH ₃)
20	124.6 (s)		124.3 (s)	
21	140.0 (d)	7.23 (s)	139.9 (d)	7.19 (s)
22	111.6 (d)	6.32 (s)	111.1 (d)	6.24 (s)
23	142.1 (d)	7.35 (br s)	142.4 (d)	7.33 (br s)
28	28.8 (q)	1.36 (s)	30.6 (q)	1.40 (s, CH ₃)
29	25.7 (q)	1.61 (s)	27.7 (q)	1.45 (s, CH ₃)
30a	34.7 (t)	2.30 (dd, 14.2, 9.1)	118.8 (t)	5.12 (s)
30b		2.97 (dd, 14.2, 3.7)		5.40 (s)
12-OAc	170.9 (s)		170.8 (s)	
	21.3 (q)	1.90 (s)	21.2 (q)	1.95 (s, CH ₃)
7-OMe	52.1 (q)	3.70 (s)	52.3 (q)	3.76 (s, CH ₃)

^a¹H NMR spectra were recorded at 400 MHz and ¹³C NMR spectrum at 100 MHz. ^b¹H NMR spectrum was recorded at 500 MHz.

compounds shared the same A, D, and E ring systems, as further confirmed by 2D NMR studies. Furthermore, the replacement of a C-30 sp² methylene signal in munronoid B^{5b} with a sp³ methylene signal (δ_{H} 2.30, 2.97; δ_{C} 34.7) in **1**, as well as the striking presence of two double bonds ($\Delta^{8(9)}$ and $\Delta^{1(10)}$) in **1** implied that the methylene group (C-30) might be the core linkage between rings A and C instead of the usual connectivity via C-9 and C-10.

(11) (a) Lidert, Z.; Taylor, D. A. H.; Thirugnanam, M. *J. Nat. Prod.* **1985**, *48*, 843–845. (b) Sarker, S. D.; Savchenko, T.; Whiting, P.; Sik, V.; Dinan, L. *Arch. Insect Biochem. Physiol.* **1997**, *35*, 211–217.

(12) (a) Zhang, Y.; Di, Y. T.; Zhang, Q.; Mu, S. Z.; Tan, C. J.; Fang, X.; He, H. P.; Li, S. L.; Hao, X. J. *Org. Lett.* **2009**, *11*, 5414–5417. (b) Zhang, Q.; Di, Y. T.; Li, C. S.; Fang, X.; Tan, C. J.; Zhang, Z.; Zhang, Y.; He, H. P.; Li, S. L.; Hao, X. J. *Org. Lett.* **2009**, *11*, 2357–2359.

(13) Frisch, M. J. et al. *Gaussian 03*, Revision D.01; Gaussian, Inc.: Wallingford, CT, 2005 (see the Supporting Information for full reference).

(14) (a) Tan, C. J.; Di, Y. T.; Wang, Y. H.; Zhang, Y.; Si, Y. K.; Zhang, Q.; Gao, S.; Hu, X. J.; Fang, X.; Li, S. F.; Hao, X. J. *Org. Lett.* **2010**, *12*, 2370–2373. (b) Wang, L.; He, H. P.; Di, Y. T.; Zhang, Y.; Hao, X. J. *Tetrahedron Lett.* **2012**, *53*, 1576–1578.

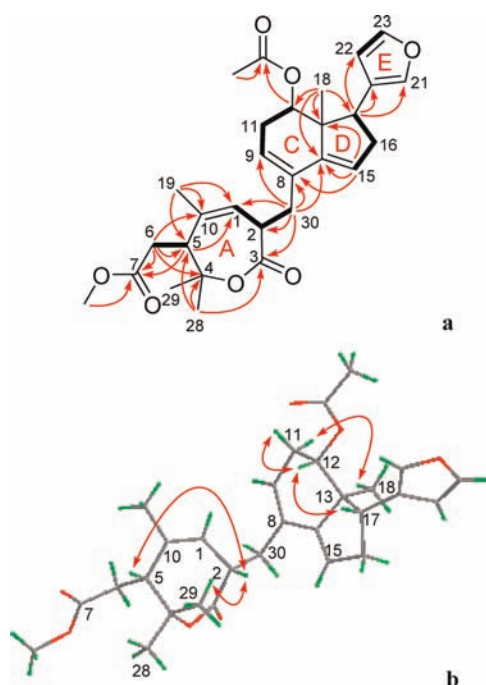


Figure 1. Key ^1H – ^1H COSY (a: —), HMBC (a: → (red)), and ROESY (b: ↔ (red)) correlations of **1**.

HMBC correlations of H_3 -19/C-1 (δ_{C} 120.2), C-5 (δ_{C} 47.8), and C-10 (δ_{C} 137.8), H-5 (δ_{H} 2.74)/C-5 and C-10 as well as the downfield-shifted hydrogen resonances of H-1 (δ_{H} 5.18) and ^1H – ^1H COSY correlations (H-5–H-6) indicated the location of a double bond between C-1 and the quaternary carbon atom C-10, which also suggested the cleavage of C-9 and C-10. The ^1H – ^1H COSY correlations (H-1–H-2–H₂-30) and the HMBC correlations of H₂-30/C-1, C-2, C-3, C-8 (δ_{C} 131.0), and C-9 (δ_{C} 125.0) confirmed that C-2 and C-8 were linked through C-30 as shown in Figure 1. HMBC correlations of H-30/C-14 (δ_{C} 147.6), H-15 (δ_{H} 5.65)/C-8, and C-14 indicated the presence of $\Delta^{14(15)}$ double bond, which was conjugated with $\Delta^{9(8)}$ double bond. Additionally, the acetoxy group was located at C-12 by the HMBC cross signal H-12/C-12-OAc, and the HMBC correlations of H₃-28/C-3 together with the downfield-shifted carbon resonances of C-3 and C-4 definitely indicated the linkage of C-3 and C-4 via an oxygen atom to form the unsaturated lactone ring A. Thus, the aforementioned data suggested a unique ring A, B-seco limonoid with a unique C-2–C-30 bond, and the planar structure of **1** was established as shown in Figure 1.

The relative stereochemistry of **1** was determined by ROESY spectrum (Figure 1b). Furthermore, the successful performance of the X-ray diffraction experiment with Cu K α radiation confirmed the proposed structure and also

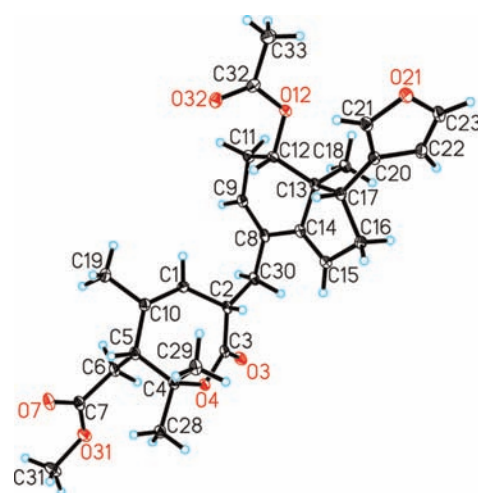


Figure 2. Single crystal X-ray structure of **1**.

allowed unambiguous assignment of the absolute configuration of **1** as drawn [Flack parameter: 0.1(2)]⁹ (Figure 2).

Aphanamixoid B (**2**)¹⁰ was obtained as colorless amorphous powder. The molecular formula was determined as C₂₉H₃₆O₈ with 12 degrees of unsaturation deduced by HRESIMS at m/z 535.2314 [$\text{M} + \text{Na}$]⁺ (calcd 535.2307, C₂₉H₃₆O₈Na). The IR absorption band at 1739 cm⁻¹ indicated the presence of ester carbonyl groups. The observation for a β -furan ring (δ_{H} 6.24, 7.19, 7.33; δ_{C} 111.1, 139.9, 142.4), a methoxy group (δ_{H} 3.76; δ_{C} 52.3), four tertiary methyl groups (δ_{H} 0.80, 1.17, 1.40, 1.45), and a characteristic exocyclic double bond (δ_{H} 5.12, 5.40) in the ^1H and ^{13}C NMR spectra of **2** strongly suggested that **2** was a prierianin-type limonoid.^{4d,11} The ^1H and ^{13}C NMR spectral data of **2** showed close similarity to those of the reported compound, munronoid A,^{5b} except for the absence of an acetyl group, as well as the presence of an additional oxygenated methine (δ_{H} 4.11). Compared with munronoid A, the observed significant downfield shifts of C-1 (δ_{C} 81.6) and C-11 (δ_{C} 79.3) together with the strong HMBC correlation (Figure 3a) connected via an oxygen atom and formed a tetrahydrofuran ring. The structure of **2** was confirmed by 2D NMR (HSQC, HMBC, ^1H – ^1H COSY, and ROESY) experiments (Figure 3).

The absolute configuration of aphanamixoid B (**2**) was assigned using the quantum chemical method. The optical rotation (OR) value of **2** was calculated using density functional theory (DFT) methods¹² in the Gaussian 03 program package.¹³ The “self-consistent reaction field” method (SCRf) was employed to perform the OR calculation of the most stable conformer of **2** in MeOH solution at the B3LYP/6-31G (d,p) level. The calculated OR value (+91.4°) for **2** is close to its experimental value (+81.8°), which suggested a reliable absolute configuration assignment for **2**. In addition, its electronic circular dichroism (ECD)¹⁴ was also calculated on the Gaussian 03 program using TD-DFT-B3LYP/6-31G(d,p) level,¹³ which showed a good agreement with those of experimentally recorded

(15) (a) Luo, S. H.; Luo, Q.; Niu, X. M.; Xie, M. J.; Zhao, X.; Schneider, B.; Gershenzon, J.; Li, S. H. *Angew. Chem., Int. Ed.* **2010**, *49*, 4471–4475. (b) Luo, S. H.; Weng, L. H.; Xie, M. J.; Li, X. N.; Hua, J.; Zhao, X.; Li, S. H. *Org. Lett.* **2011**, *13*, 1864–1867.

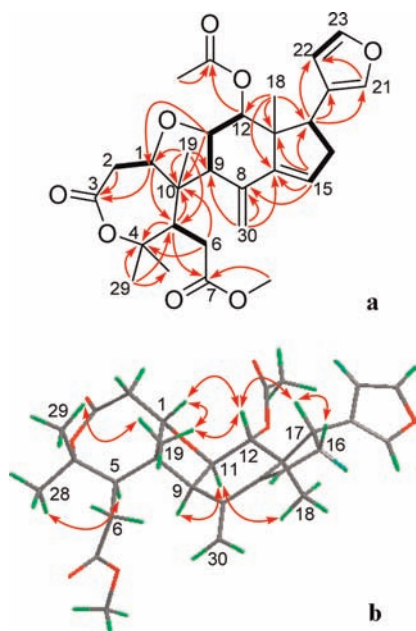


Figure 3. Key ^1H - ^1H COSY (a: \rightarrow), HMBC (a: \rightarrow (red)), and ROESY (b: \leftrightarrow (red)) correlations of **2**.

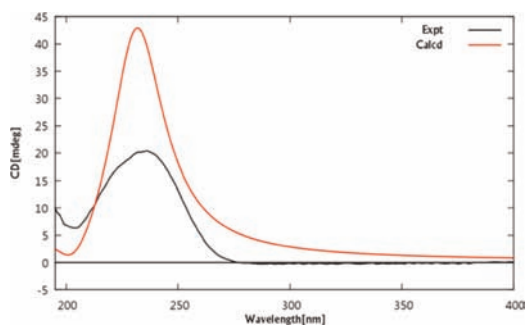


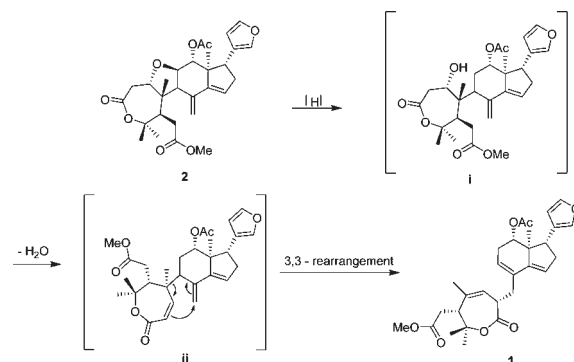
Figure 4. Comparison of the experimental CD and calculated CD spectrum of **2**.

CD spectrum (Figure 4). Thus, the absolute configuration of **2** was unambiguously assigned as depicted.

The biogenetic origin of aphanamixoid A (**1**) might be derived from aphanamixoid B (**2**). The cleavage of ether

linkage followed by reduction of **2** formed **i**, and then dehydration to yield the key intermediate **ii**, which produced **1** by means of 3,3-rearrangement, as shown in Scheme 1.

Scheme 1. Plausible Biosynthetic Pathway for **1**



The antifeedant activity of aphanamixoid A (**1**) against the larvae of two generalist insects, beet armyworm (*Spodoptera exigua*) and cotton bollworm (*Helicoverpa armigera*), were evaluated.¹⁵ The compound **1** exhibited a potent antifeedant activity with an EC_{50} value of 0.052 and 0.015 $\mu\text{mol}/\text{cm}^2$, respectively. The results suggested a potent defensive role of **1** against herbivore enemies, while aphanamixoid B (**2**) showed moderate antifeedant activity against *S. exigua* at 2000 ppm with AFI of 17%.

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Supporting Information Available. 1D and 2D NMR spectra of **1** and **2**, experimental procedures, plant material, bioactivity assay, the X-ray crystallographic data for **1**, and computational calculations for **2**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

The authors declare no competing financial interest.